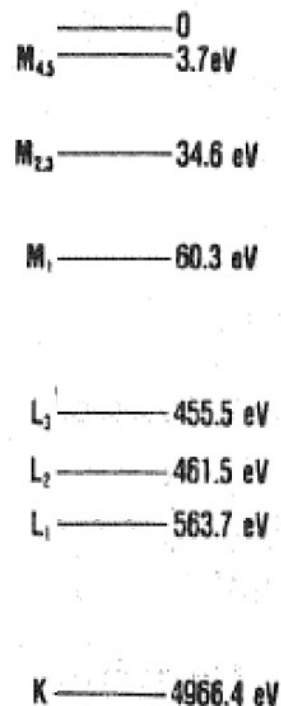


**Problem 1 (10 points). Select True or false**

- |  |           |       |
|--|-----------|-------|
| 1. Auger electrons cannot be produced in XPS measurements                            | . . . . . | T_ F_ |
| 2. Photoelectrons cannot be produced in XRF measurements                             | . . . . . | T_ F_ |
| 3. X-ray fluorescence cannot be produced in XPS measurements                         | . . . . . | T_ F_ |
| 4. Auger emission can only be achieved in ultra high vacuum                          | . . . . . | T_ F_ |
| 5. $dN(E)/dE$ is used to study XPS spectra.  | . . . . . | T_ F_ |
| 6. Sample charging is never a problem in XPS.  | . . . . . | T_ F_ |
| 7. Elements with high atomic number (>50) most often produce KLL Auger electrons.    | . . . . . | T_ F_ |
| 8. The background of XRF is composed mostly of secondary electrons.                  | . . . . . | T_ F_ |
| 9. The kinetic energy of Auger electrons depends on the energy of the electron beam. | . . . . . | T_ F_ |
| 10. Electron beams used in AES typically have energies between 1 and 2 keV.          | . . . . . | T_ F_ |

**Problem 2 (10 points).** An atom of titanium undergoes an Auger process when hit by an external electron of kinetic energy of 5030 eV. Describe a possible sequence of steps of the Auger process including the kinetic energies of all emitted particles and of all energy level transitions.



**Problem 3 (10 points).** The diagram shows the number of electrons as a function of their kinetic energy obtained in an XPS measurement of Pd metal performed with Mg  $K\alpha$  X-rays of energy 1253.6 eV. Neglecting any work function of the spectrometer, what can you say about the binding energies of the three major peaks shown?

